

This article was downloaded by: [Tomsk State University of Control Systems and Radio]

On: 21 February 2013, At: 12:40

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954

Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/gmcl16>

### High Pressure Investigations of the Phase Transitions in Cholesteryl Pentanoate and Cholesteryl Hexanoate

P. Pollmann<sup>a</sup> & B. Wiege<sup>a</sup>

<sup>a</sup> Department of Physical Chemistry, University of Paderborn, 4790, Paderborn, Germany

Version of record first published: 15 Sep 2010.

To cite this article: P. Pollmann & B. Wiege (1982): High Pressure Investigations of the Phase Transitions in Cholesteryl Pentanoate and Cholesteryl Hexanoate, Molecular Crystals and Liquid Crystals, 72:9-10, 271-276

To link to this article: <http://dx.doi.org/10.1080/01406568208084720>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable

for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## HIGH PRESSURE INVESTIGATIONS OF THE PHASE TRANSITIONS IN CHOLESTERYL PENTANOATE AND CHOLESTERYL HEXANOATE

P. POLLMANN and B. WIEGE  
 Department of Physical Chemistry, University of  
 Paderborn, 4790 Paderborn, Germany

(Submitted for Publication January 4, 1982)

**Abstract:** At high pressures too cholesteryl n-pentanoate (CH-5) and cholesteryl n-hexanoate (CH-6) exhibit only a cholesteric liquid crystalline phase. Our light reflection measurements up to 2600 bars yield no indication of a pressure induced smectic phase which was suggested by Tikhomirova and Ginzberg<sup>1</sup> for CH-5 at about 850 bars and 120°C. The obtained pressure( $p_t$ )-temperature ( $T_t$ ) phase diagrams show the expected widening of the cholesteric range with  $p_t$  and  $T_t$ . In the case of CH-5 the marked crystallization curve besides a melting curve could be explained by the transformation of the cholesteric into a metastable crystalline solid phase.

**Introduction** In the homologous series of the cholesteryl n-alkanoates the lower homologues exhibit only cholesteric mesophases at atmospheric pressure, but from the heptanoate both smectic A and cholesteric mesophases are observed. Therefore, we suppose, the pressure induced phase of cholesteryl pentanoate (CH-5) found by the Russian workers Tikhomirova and Ginzberg<sup>1</sup> is assumed by them to be a smectic. Because our laboratory is equipped with an optical high pressure apparatus for sensitive detection of smectic phases and the available pressure( $p_t$ )-temperature( $T_t^*$ ) phase diagrams of CH-5<sup>1,3,4</sup> differ very much from one another, we have re-investigated the  $p_t$ - $T_t$  phase behaviour of this alkanoate.

Compared with the CH-5 pressure induced smectic behaviour is rather to be expected for the hexanoate (CH-6) (see above). On that account the phase behaviour of this cholesteryl ester was studied too.

**Results and discussion** A smectic A phase ( $S_A$ ) in addition to a cholesteric phase (CH) can be established by the divergence of the maximum wavelength of the selective light reflection  $\lambda_R$  of the latter phase near the phase transition CH/ $S_A$ . This effect is shown in figure 1 for cholesteryl nonanoate at atmospheric pressure. An analogous behaviour of  $\lambda_R$  occurs when the cholesteric compound is pressurized to the CH/ $S_A$  transition

\*) Index t means transition

at constant temperature.

The supposed induced smectic mesophase of CH-5 was obtained by Tikhomirova and Ginzberg<sup>1</sup> at about 850 bars and 120°C. No divergence of  $\lambda_R$ , however, was found by us in our experimental range up to 2600 bars and 169.4°C.

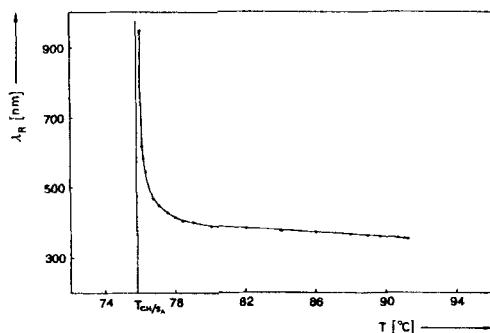


FIGURE 1 Temperature ( $T$ ) dependence of the maximum wavelength of the selective light reflection  $\lambda_R$  for cholesteryl nonanoate at atmospheric pressure  
 $T_{CH/S_A}$  = phase transition temperature cholesteric (CH)/smectic A ( $S_A$ )

A slight red shift of  $\lambda_R$  with pressure was only observed at different constant temperatures as shown in figure 2 for 169.4°C.

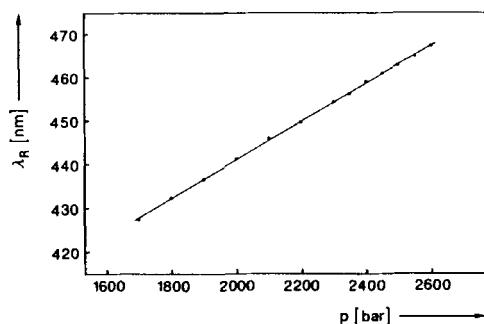


FIGURE 2 Pressure ( $p$ ) dependence of the maximum wavelength of the selective light reflection  $\lambda_R$  for cholesteryl pentanoate at 169.4°C

The upper pressure limit of the isotherm is the phase transition pressure cholesteric (CH)/crystalline (C). For CH-6 too in the range 1-1225 bars and 93.62 - 136.02°C no divergence of  $\lambda_R$  was observed. The transition points CH/C as well as cholesteric (CH)/isotropic (I) which was also determined are indicated by the sudden disappearance of the light reflection of the cholesteric phase. Because CH-5 as well as CH-6 exhibit a blue phase (BP), upon closer consideration the transition CH/I must be replaced by CH/BP and BP/I respectively. The range of the BP, however, is only about 0.5 degrees, and therefore ignored here and in the following.\*)

Not shown in this paper are the numerous isotherms of  $\lambda_R$  versus pressure necessary for the determination of the  $p_t$ - $T_t$  phase diagrams of CH-5 and CH-6. For CH-5, however, the slopes of these isotherms  $(\partial\lambda_R/\partial p)_T$  are given in figure 3 as functions of temperature.

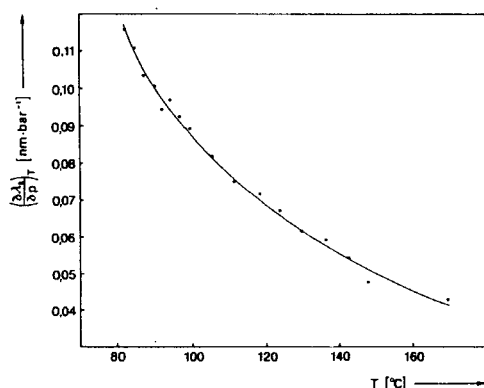


FIGURE 3 Temperature dependence of the first derivative of the  $\lambda_R(p)$  values  $(\partial\lambda_R/\partial p)_T$  for cholesteryl pentanoate. Meaning of letters as in figure 1 and 2

The observed decrease of  $(\partial\lambda_R/\partial p)_T$  with temperature is an additional argument against a pressure induced smectic phase.  $(\partial\lambda_R/\partial p)_T$  could be easily determined, because the isotherms are all linear in the respective upper pressure ranges. The decrease of  $(\partial\lambda_R/\partial p)_T$  with increasing temperature cannot according to

$$\lambda_R = \bar{n} \cdot z \quad (1)$$

$\bar{n}$  = average refractive index of the cholesteric phase  
 $z$  = pitch of the cholesteric helix

\*)  $p_t$ - $T_t$  phase behaviour of the BP see<sup>5</sup>

be attributed to the pressure ( $p$ )-temperature ( $T$ ) behaviour of  $\bar{n}$  and the density, which has a direct influence on  $z$ . Both effects have been estimated. In the case of  $\bar{n}$  we have made use of the measurements of Horn<sup>6</sup> who observed the  $p$ - $T$  dependence of the two refractive indices of a nematic up to 2000 bars. For  $z$  we determined the direct dependence on the density from the  $p$ - $V$ - $T$  data of CH-5<sup>4</sup> assuming an isotropic compressibility. The influences of  $\bar{n}$  as well as of  $z$  and of the density respectively on  $\lambda_R$  are of no importance and moreover cancel out. Hence there remains a significant decrease of the pressure dependence of  $z$  (according to eq. (1)) with increasing temperature which means a decrease of that of the twist angle ( $\propto z^{-1}$ ).

In figure 4 and 5 the  $p_t$ - $T_t$  phase diagrams of CH-5 and CH-6 based upon our light reflection measurements are presented. The diagrams show the cholesteric liquid crystal (CH) between the crystalline solid (C) and the isotropic liquid phase (I) (The very small blue phase region has been neglected). In both cases the cholesteric range is widened by pressure. Furthermore, a crystallization (a) and a melting curve (b) can be clearly distinguished with remarkable small deviations of the experimental points from CH-5 crystallization curve (see fig. 4, curve a).

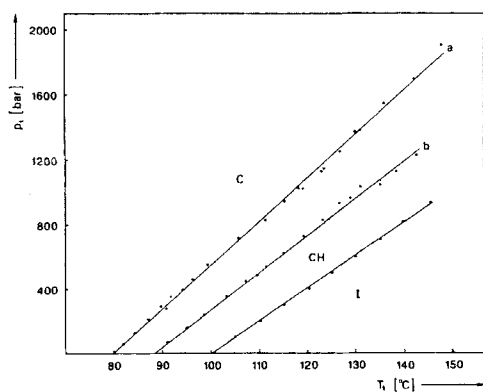


FIGURE 4  $p_t$ - $T_t$  phase diagram of cholesteryl pentanoate  
(C = crystalline, CH = cholesteric, I = isotropic)  
 $p_t$ ,  $T_t$  = phase transition pressure and temperature respectively  
a = crystallization curve, b = melting curve

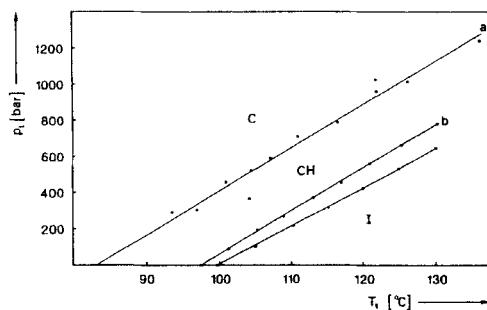


FIGURE 5  $p_t$ - $T_t$  phase diagram of cholesteryl hexanoate  
(C = crystalline, CH = cholesteric, I = isotropic)  
 $p_t$ ,  $T_t$  = phase transition pressure and temperature respectively  
a = crystallization curve, b = melting curve

For this curve a of CH-5, contrary to CH-6, a transformation of the cholesteric into a metastable crystalline phase could be established by X-ray measurements<sup>7</sup>.

The accordance of our  $p_t$ - $T_t$  phase diagrams with those of two Russian groups<sup>1,4,8</sup> is low. While our data of the phase transition temperatures at atmospheric pressure agree well with those given in the literature, those of<sup>1,4,8</sup> do not. According to<sup>4,8</sup> CH-5 and CH-6 should show smectic behaviour even at atmospheric pressure. However as outlined above, no smectic behaviour could be observed by us at higher pressures too. The slopes ( $dp_t/dT_t$ ) for the melting and clearing curve of CH-5 and CH-6 in this work, however, differ very little from those obtained by Robberecht<sup>3</sup> who investigated these compounds already in 1938 up to 760 bars.

#### Acknowledgement

This work has been supported by Deutsche Forschungsgemeinschaft and Fonds der Chemischen Industrie.

References

1. N.A. Tikhomirova and A.V. Ginzberg, Kristallografiya 22, 155 (1977)
2. P. Pollmann, J. Phys. E 7, 490 (1974)
3. J. Robberecht, Bull. Soc. chim. Belgique 47, 597 (1938)
4. V. Ya. Baskakov and V.K. Semenchenko, JETP Lett. 17, 414 (1973)
5. P. Pollmann and G. Scherer, High Temp.-High Press. 12, 103 (1980)
6. R.G. Horn, J. Physique 39, 167 (1978)
7. E.H. Pape, University of Paderborn, unpublished results
8. V.M. Byankin, V.K. Semenchenko and V.Ya. Baskakov, Russ. J. Phys. Chem. 48, 727 (1974)